## Mathematical foundations of modern quantum simulations



Fabian M. Faulstich

Department of Mathematics, Rensselaer Polytechnic Institute, Troy, NY

## What are quantum simulations?



Stability of matter


Color of gold


Superconductivity


Solid state physics

## Towards a mathematical formulation



## Towards a mathematical formulation



Absorbtion of Hydrogen atom


## Towards a mathematical formulation

Two energetic contributions:

1. Kinetic energy: $h_{\text {kin }}=-\frac{1}{2} \Delta$
2. Coulomb energy: $V_{\text {coul }}=-\frac{1}{|r-R|}$


## Towards a mathematical formulation

Absorbtion of Hydrogen atom

Two energetic contributions:

1. Kinetic energy: $h_{\text {kin }}=-\frac{1}{2} \Delta$
2. Coulomb energy: $V_{\text {coul }}=-\frac{1}{|r-R|}$


Birth of quantum chemistry ${ }^{a}$ :

$$
\Delta \psi+\frac{2 m}{K^{2}}\left(E+\frac{e^{2}}{r}\right) \psi=0
$$

${ }^{2}$ Schrödinger, Annalen der physik (1926)

## Schrödinger equation

The goal is to solve

$$
H \Psi=E \Psi
$$

where

$$
H=-\frac{1}{2} \sum_{i=1}^{N} \Delta_{r_{i}}-\sum_{i=1}^{N} \sum_{j=1}^{N_{\mathrm{nuc}}} \frac{Z_{j}}{\left|r_{i}-R_{j}\right|}+\sum_{i=1}^{N} \sum_{j>i}^{N} \frac{1}{\left|r_{i}-r_{j}\right|}
$$

and $\Psi$ is a function defined on $\mathbb{R}^{3 N}$ for $N$ electrons.
$r_{i}$ position of the ith electron
$R_{j}, Z_{j}$ position (fixed) and charge of the $\mathrm{j} t \mathrm{~h}$ nucleus

## What is the problem?

Discretization scales exponentially in the number of electrons $N$
${ }^{1}$ The estimated number of particles in the universe is $\sim 10^{80}$

## What is the problem?

Discretization scales exponentially in the number of electrons $N$

Example: carbon dioxide $\left(\mathrm{CO}_{2}\right)$

Oxygen has 8 electrons
Carbon has 6 electrons
In total: $N=2 * 8+6=22$
${ }^{1}$ The estimated number of particles in the universe is $\sim 10^{80}$

## What is the problem?

Discretization scales exponentially in the number of electrons $N$

Example: carbon dioxide $\left(\mathrm{CO}_{2}\right)$

Oxygen has 8 electrons
Carbon has 6 electrons
In total: $N=2 * 8+6=22$

Naïve grid with 10 points per axis yields $10^{3 \cdot 22}=10^{66}$ grid points ${ }^{1}$
${ }^{1}$ The estimated number of particles in the universe is $\sim 10^{80}$

## What is the problem?

Discretization scales exponentially in the number of electrons $N$

Example: carbon dioxide $\left(\mathrm{CO}_{2}\right)$

Oxygen has 8 electrons
Carbon has 6 electrons
In total: $N=2 * 8+6=22$

Naïve grid with 10 points per axis yields $10^{3.22}=10^{66}$ grid points ${ }^{1}$

$$
\Rightarrow \text { Approximation is key! }
$$

${ }^{1}$ The estimated number of particles in the universe is $\sim 10^{80}$

## Quantum chemical methods



## Strongly correlated quantum chemistry

## Strongly correlated quantum chemistry



This happens at various scales


Small molecules


Transition metal complexes


Twisted bilayer graphene

## Strongly correlated quantum chemistry



This happens at various scales


Small molecules
Transition metal complexes


Strongly correlated systems are very difficult to compute

## Strongly correlated quantum chemistry



This happens at various scales


Small molecules
Transition metal complexes


Twisted bilayer graphene

Strongly correlated systems are very difficult to compute Important failure modes for existing methods
$\Rightarrow$ The research frontier of computational chemistry lies in strongly correlated systems

## Roadmap



1. Quantum mechanics of the hydrogen atom
2. The hydrogen molecule and the Hartree-Fock method
3. Second quantization I
4. Second quantization II
5. Second quantization III
6. The linear combination of atomic orbitals ansatz
7. Gaussian-type atomic orbitals and high-dimensional integration
8. Size consistency and the coupled cluster ansatz
9. Truncating the coupled cluster ansatz and its working equations
10. Finding one root to the coupled cluster equations
11. Finding all roots to the coupled cluster equations

## Quantum mechanics of the hydrogen atom

Analytic solution to the eigenvalue problem

$$
\left(\frac{1}{2} \Delta_{r}+\frac{1}{|r-R|}\right) \Psi(r)=-E \Psi(r)
$$

3D partial differential equation!

$Y_{3}^{0}=5 \cos ^{3} \theta-3 \cos \theta$

${ }^{c} Y_{3}^{1}=\left(5 \cos ^{2} \theta-1\right) \sin \theta \cos \phi$


## The hydrogen molecule and the Hartree-Fock method

The Hamiltonian reads

$$
\begin{aligned}
H= & -\frac{1}{2} \Delta_{r_{1}}-\frac{1}{\left|r_{1}-R_{1}\right|}-\frac{1}{2} \Delta_{r_{2}}-\frac{1}{\left|r_{2}-R_{2}\right|} \\
& -\frac{1}{\left|r_{2}-R_{1}\right|}-\frac{1}{\left|r_{1}-R_{2}\right|}+\frac{1}{\left|r_{1}-r_{2}\right|}
\end{aligned}
$$

we want to solve

$$
H \Psi\left(r_{1}, r_{2}\right)=E \Psi\left(r_{1}, r_{2}\right)
$$

$\Rightarrow$ Not possible!


## Second quantization I:

Fock space, creation and annihiliation operators, CAR algebra

High-dimensional PDE's are hard!

- Can we think of matrices instead?
- Can we manipulate these matrices fast?


## Second Quantization



## Second quantization II:

Excitation operators, CCR algebra, nilpotent Lie algebra

The second quantization shows its teeth


## Second quantization III:

Slater-Condon rules, Wick's theorem, Hartree-Fock revisited

Let's bring out the big guns...


## The linear combination of atomic orbitals ansatz

Atomic orbitals


Molecular orbitals


$$
\frac{Y}{\eta} x
$$



## Gaussian-type atomic orbitals and high-dimensional integration

We will face the following integral evaluation:

$$
v_{p, q, r, s}=\int_{X \times X} \frac{\xi_{p}^{*}\left(x_{1}\right) \xi_{q}\left(x_{1}\right) \xi_{r}^{*}\left(x_{2}\right) \xi_{s}\left(x_{2}\right)}{\left|r_{1}-r_{2}\right|} d \lambda(x) d \lambda(x)
$$



## Size consistency and the coupled cluster ansatz

What do we really care about?


## Truncating the coupled cluster ansatz and its working equations

$$
\begin{aligned}
& E(t)=\left\langle\Psi_{0}, H \Psi_{0}\right\rangle+\sum_{I A} f_{I A} t_{I}^{A}+\frac{1}{4} \sum_{I J A B}\langle I J \| A B\rangle t_{I J}^{A B}+\frac{1}{2} \sum_{I J A B}\langle I J \| A B\rangle t_{I}^{A} t_{I}^{B}, \\
& f(t)_{I}^{A}=f_{I A}+\sum_{C} f_{A C} t_{I}^{C}-\sum_{K} f_{K I} t_{K}^{A}+\sum_{K C}\langle K A \| C I\rangle t_{C}^{K}+\sum_{K C} f_{K C} t_{I K}^{A C} \\
& +\frac{1}{2} \sum_{K C D}\langle K A \| C D\rangle t_{K I}^{C D}-\frac{1}{2} \sum_{K L C}\langle K L \| C I\rangle t_{K L}^{C A}-\sum_{K C} f_{K C} t_{I}^{C} t_{K}^{A}-\sum_{K L C}\langle K L \| C I\rangle t_{K}^{C} t_{L}^{A} \\
& +\sum_{K C D}\langle K A \| C D\rangle t_{K}^{C} t_{I}^{D}-\sum_{K L C D}\langle K L \| C D\rangle t_{K}^{C} t_{I}^{D} t_{L}^{A}+\sum_{K L C D}\langle K L \| C D\rangle t_{C}^{K} t_{L I}^{D A} \\
& -\frac{1}{2} \sum_{K L C D}\langle K L \| C D\rangle t_{K I}^{C D} t_{L}^{A}-\frac{1}{2} \sum_{K L C D}\langle K L \| C D) t_{K}^{C A} t_{I}^{D} \\
& f(t)_{I J}^{A B}=\langle I J \| A B\rangle+\sum_{C}\left(f_{B C} t_{I J}^{A C}-f_{A C} t_{I J}^{B C}\right)-\sum_{K}\left(f_{K J} t_{I K}^{A B}-f_{K I}{ }^{t} A B\right) \\
& +\frac{1}{2} \sum_{K L}\langle K L \| I J\rangle t t_{K L}^{A B}+\frac{1}{2} \sum_{C D}(A B \| C D\rangle t_{I J}^{C D}+P(I J) P(A B) \sum_{K C}\langle K B \| C J\rangle t_{I K}^{A C} \\
& +P(I J) \sum_{C}\langle A B \| C J\rangle t_{I}^{C}-P(A B) \sum_{K}\langle K B \| I J\rangle t_{A}^{K} \\
& +\frac{1}{2} P(I J) P(A B) \sum_{K L C D}\langle K L \| C D\rangle t_{I K}^{A C} t_{L J}^{D B}+\frac{1}{4} \sum_{K L C D}\langle K L \| C D\rangle t_{I J}^{C D} t_{K L}^{A B} \\
& +\frac{1}{2} P(A B) \sum_{K L C D}\langle K L \| C D\rangle t_{I J}^{A C} t_{K L}^{B D}-\frac{1}{2} P(I J) \sum_{K L C D}\langle K L \| C D) t_{I K}^{A B} t_{J L} \\
& +\frac{1}{2} P(A B) \sum_{K L}\langle K L \| I J\rangle t_{K}^{A} t_{L}^{B}+\frac{1}{2} P(I J) \sum_{C D}\langle A B \| C D\rangle t_{I}^{C} t_{J}^{D} \\
& -P(I J) P(A B) \sum_{K C}\langle K B \| I C) t_{K}^{A} t_{J}^{C}+P(A B) \sum_{K C} f_{K C} t_{K}^{A} t_{I J}^{B C} \\
& +P(I J) \sum_{K C} f_{K C} t_{I}^{C} t_{J K}^{A B}-P(I J) \sum_{K L C}(K L \| C I) t_{K}^{C} t_{L J}^{A B} \\
& +P(A B) \sum_{K C D}\langle K A \| C D\rangle t_{K}^{C} t_{I J}^{D B}+P(I J) P(A B) \sum_{K C D}\langle A K \| D C\rangle t_{I}^{D} t_{J K}^{B C} \\
& +P(I J) P(A B) \sum_{K L C}\langle K L \| I C\rangle t_{L}^{A} t_{J K}^{B C}+\frac{1}{2} P(I J) \sum_{K L C}\langle K L \| C J\rangle t_{I}^{C} t_{K L}^{A B} \\
& -\frac{1}{2} P(A B) \sum_{K C D}\langle K B \| C D\rangle t_{K}^{A} t_{I J}^{C D}+\frac{1}{2} P(I, J) P(A B) \sum_{K L C}\langle K B \| C D) t_{I}^{t^{t}}{ }_{K}^{A} t_{J}^{D} \\
& +\frac{1}{2} P(I J) P(A B) \sum_{K L C}\langle K L \| C J\rangle t_{I}^{C} t_{K}^{A} t_{L}^{B}-P(I J) \sum_{K L C D}\langle K L \| C D\rangle t_{K}^{C} t_{I}^{D} t_{L}^{A B} \\
& -P(A B) \sum_{K L C D}\langle K L \| C D\rangle t_{K}^{C} t_{L}^{A} t_{I J}^{D B}-\frac{1}{4} P(I J) \sum_{K L C D}\langle K L \| C D\rangle t_{I}^{C} t_{J}^{D} t_{K}^{A B} \\
& +\frac{1}{4} P(A B) \sum_{K L C D}\langle K L \| C D\rangle t_{K}^{A} t_{L}^{B} t_{I J}^{C D}+P(I J) P(A B) \sum_{K L C D}\langle K L \| C D) t_{I}^{G} t_{L}^{B} t_{K}^{A D} \\
& +\frac{1}{4} P(I J) P(A B) \sum_{K L C D}\langle K L \| C D\rangle t_{I}^{C} t_{K}^{A} t_{J}^{D} t_{L}^{B}
\end{aligned}
$$

Finding one root to the coupled cluster equations


## Finding all roots to the coupled cluster equations

Seeking:

$$
f(\mathrm{t})=\left[\begin{array}{c}
f_{1}(\mathrm{t}) \\
\vdots \\
f_{m}(\mathrm{t})
\end{array}\right]=\left[\begin{array}{c}
f_{1}\left(t_{1}, \ldots, t_{m}\right) \\
\vdots \\
f_{m}\left(t_{1}, \ldots, t_{m}\right)
\end{array}\right]=0
$$

Find a start system $g$, and then continuously deform $g$ into $f$

Davidenko differential equation:

$$
\frac{\partial}{\partial \mathrm{t}} H(\mathrm{t}, \lambda)\left(\frac{\mathrm{d}}{\mathrm{~d} \lambda} \mathrm{t}(\lambda)\right)+\frac{\partial}{\partial \lambda} H(\mathrm{t}, \lambda)=0, \quad \mathrm{t}(1)=\mathrm{s}_{0},
$$

Finding all roots to the coupled cluster equations
Seeking:

$$
f(\mathrm{t})=\left[\begin{array}{c}
f_{1}(\mathrm{t}) \\
\vdots \\
f_{m}(\mathrm{t})
\end{array}\right]=\left[\begin{array}{c}
f_{1}\left(t_{1}, \ldots, t_{m}\right) \\
\vdots \\
f_{m}\left(t_{1}, \ldots, t_{m}\right)
\end{array}\right]=0
$$



